Ambient concentrations and total deposition of inorganic sulfur, inorganic nitrogen and base cations in the Athabasca Oil Sands Region

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HIGHLIGHTS
- Atmospheric concentration and deposition data from 2000 to 2017 were analyzed.
- Deposition of S, N and base cations was estimated for boreal jack pine sites.
- Highest ambient concentrations and deposition occurred near oil sand operations.
- Potential acid input exhibited a complex pattern dependent on local sources.
- Significant downward trends were observed for SO2, less so for NO2 and PM.

GRAPHICAL ABSTRACT

ABSTRACT
Trace gas, particulate matter and deposition data collected in the Athabasca Oil Sands Region (AOSR) from 2000 to 2017 were evaluated as part of a broad scientific programmatic review. Results showed significant spatial patterns and temporal trends across the region. Concentrations of reactive gases were highest near the center of surface oil sands production operations and decreased towards the edges of the monitoring domain by factors of 8, 20, 4 and 3 for SO2, NO2, HNO3 and NH3, respectively. 18 of 30 sites showed statistically significant (p < 0.05) negative trends in SO2 concentrations suggesting an ~40% decrease since 2000. In contrast, only 2 of 30 sites showed statistically significant temporal trends (1 positive, 1 negative) for NO2. NH3 data showed (i) intermittent wildfire impacts, and (ii) high seasonality, with low concentrations during winter and significantly higher values during the summer. PM10 measurements were more limited, but also showed significant spatio-temporal variability. Comparison of PM10 and PM2.5 data showed that >80% of SO4 2− was in the PM2.5 fraction, while > 60% of Ca2+, Mg2+, Na+ and Cl− were in the PM10 fraction. Ion balances of both PM10 and PM2.5 contained cation excesses at near-field oil sand sites, but PM2.5 samples at forest health sites >20 km from surface production locations contained anion excesses. Monthly average concentrations of PM10 ions showed peak Ca2+ during March-April to November, but peak SO4 2−, NH4+ and NO3− from November-March. Deposition estimates showed rapid declines as a function of distance to oil sand operations. Estimated total N and total S deposition to forest health monitoring sites ranged from 2.0 to 5.7 kg ha−1 a−1 and 2.1–14.0 kg ha−1 a−1, respectively. Potential acid input (PAI) ranged from −0.46 to 0.79 keq ha−1 a−1.
1. Introduction

The Athabasca Oil Sands Region (AOSR) in northeastern Alberta, Canada contains the world’s third largest proven oil reserve at 166 billion barrels. Approximately 20% of the reserve is considered mineable (i.e., within 70 m of the surface) and the other 80% requires in situ approaches (e.g., steam-assisted gravity drainage) for extraction. Commercial development of the oil sands began in the late 1960s with surface mining operations between Fort McKay and Fort McMurray (Foster et al., 2019). Total production from oil sands operations increased roughly 4-fold from 2000 to 2015 (6.1 × 10^5 barrels per day (bbl/day) to 2.4 × 10^6 bbl/day) and is projected to increase another 60% (3.9 × 10^6 bbl/day) by 2030 (Oil Sands Magazine, 2016).

Oil sand operations are the main source of air emissions in the AOSR. Heavy duty diesel shovels and trucks are major sources of NOx (i.e., NO + NO2), fine particulate matter and black carbon. Industrial operations for extracting and upgrading bitumen (raw oil sand) to crude oil are major sources of SO2. Fugitive dust from mining operations, mine haul roads, quarries, highways and storage piles are significant sources of trace elements, base cations (BC), polycyclic aromatic hydrocarbons (PAHs) and polycyclic aromatic compounds (PACs) (Zhang et al., 2016; Landis et al., 2019). Fugitive dust emissions from these sources are poorly understood, both in terms of particle size distribution and total mass. NOx emissions in the AOSR were approximately 80 and 110 kilotons (kton), respectively, in 2000 and 2015 and are projected to increase to about 180 kton by 2030. In contrast, SO2 emissions declined from 95 kton in 2000 to 70 kton in 2015 and are projected to increase to about 110 kton by 2030 (Foster et al., 2019).

Surrounding the oil sand operations are large areas of boreal forest growing on soils with low buffering capacity and lakes with low acid neutralizing capacity. In parallel with development, concern has grown about the effects of emissions from oil sand operations on potentially sensitive terrestrial and aquatic ecosystems. Substantial work has been done in the AOSR to understand sensitivity of soils, vegetation and lakes to these stressors (Foster et al., 2001; Percy et al., 2013; Macdonald, 2015), and to develop emission inventories (Zhang et al., 2018; Qui et al., 2018), deterministic models (Davies, 2012; Vijayaraghavan et al., 2016; Makar et al., 2018) and receptor models (Landis et al., 2012, 2017a, 2019) to estimate or predict deposition/exposure and to associate sources with sinks. Numerous studies and surveys have been conducted to document contaminant concentrations and trends in air (Hsu, 2013; Hsu et al., 2016; Davidson and Spink, 2018; Bari et al., 2016; Percy et al., 2012), snowpack (Murray, 1981; Guéguen et al., 2016), wet deposition (Barrie and Kovalick, 1980; Sandhu and Blower, 1986; Lynam et al., 2015), rivers and lakes (Kelly et al., 2010), lichens (Addison and Puckett, 1980; Landis et al., 2012, 2019), sphagnum moss (Suttyk et al., 2014, 2016) and other components of boreal ecosystems. In general, these studies have shown that emissions from the oil sands are detectable in various media (air, snowpack, lichen, moss) out to a distance of 20–50 km.

In the late 1990s, the Wood Buffalo Environmental Association (WBEA) initiated several programs for long-term monitoring of air quality, deposition and forest health in the AOSR (Foster et al., 2019). These were motivated by growing concerns about acidification and eutrophication or fertilization of surrounding ecosystems. The WBEA Terrestrial Environmental Effects Monitoring (TEEM) program established a region-wide passive sampling network in 1998–99 to monitor above-canopy concentrations of SO2, NO2, O3, HNO3 and NH3 at Forest Health Monitoring (FHM) sites (Foster et al., 2019). TEEM deployed a second network in 2008 to measure bulk and throughfall deposition of inorganic acids and base cations at FHM sites, and a third network in 2013–17 to measure above canopy concentrations of gases and PM2.5 composition at several solar-powered FHM sites. Taken together, these networks provide a relatively dense array of measurements for examining patterns and trends of deposition and air quality.

This paper presents and analyzes TEEM air and deposition data collected from 2000 through 2017, a period of rapid development in the oil sands. Observational data are used to examine temporal trends in gaseous and particulate sulfur (S), gaseous and particulate nitrogen (N) and BC across the AOSR using parametric and non-parametric statistics. TEEM data are also used to estimate total deposition of S, N and BC at TEEM FHM sites. Results are then compared with other monitoring networks in Canada, the U.S., UK and Europe. We also calculate potential acid input (PAI) for the TEEM FHM sites, referenced to the TEEM forest health survey conducted in 2011–12. PAI is a useful indicator of the relative deposition of total acids and total bases (i.e., net acidity or alkalinity) to forest ecosystems. PAI is also important from a regulatory perspective, because critical loads for PAI have been established for the Province of Alberta (Government of Alberta, 2012). Our approach for estimating PAI uses observational data at specific FHM sites, while the regulatory approach uses modelled data for discrete grid cells. Although these two approaches are not directly comparable, our findings are relevant because they shed light on deposition to one of the most sensitive receptors in the AOSR (i.e., jack pine stands).

2. Methods

Fig. 1 shows the region of interest and the sites that provided measurement data or served as receptors for deposition calculations. FHM sites have 4-digit identifiers and community or industrial sites have 4-letter identifiers. Information and associated measurements for TEEM FHM sites are listed in Appendix Tables A.1 and A.2.

2.1. Ambient air concentrations

2.1.1. Passive measurements of gaseous SO2, NO2, HNO3, and NH3

Passive samplers were used to measure sulfur dioxide (SO2), nitrogen dioxide (NO2), nitric acid (HNO3) and ammonia (NH3) at community, industrial and FHM sites (see Appendix Table A.1). SO2 and NO2 were measured with samplers developed by Tang et al. (1997 and 1999); HNO3 and NH3 were measured with Ogawa (Pompano Beach, FL) style samplers (Hsu, 2013).

The passive sampling network included 11–13 sites from 2000 to 2007, 23–28 sites from 2008 to 2012, and 30–35 sites from 2013 to 2017 (see Foster et al., 2019 for rationale and details). At FHM sites, samplers were exposed on towers several meters (m) above the forest canopy (roughly 25 m above ground level, agl). At non-FHM sites, samplers were exposed in forest clearings, roughly 5 m agl. Duplicate samples were collected at roughly half of the...
sites, and multiple travel blanks were collected with each sample batch. Exposure times were approximately 1-month during the warm season (April–September) and 2-months during the cold season (October–March).

Samples and blanks were extracted in >18.2 MΩ cm deionized water and analyzed for sulfate (SO₄²⁻), nitrite (NO₂⁻), nitrate (NO₃⁻) and ammonium (NH₄⁺) using a Dionex (Sunnyvale, CA) Model 300 ion chromatograph (IC). Loadings of SO₄²⁻, NO₂⁻, NO₃⁻ and NH₄⁺ were converted to SO₂, NO₂, HNO₃ and NH₃, respectively, based on exposure time and an empirically derived effective flow rate that was a function of ambient temperature, wind speed and relative humidity (Hsu, 2013). Detection limits (LDs) were 0.1 ppb for SO₂, NO₂, NH₃, and 0.02 ppb for HNO₃.

2.1.2. Filter-based measurements of PM₂.₅ and PM₁₀

PM (particulate matter) samples for chemical analysis were collected at four community sites and three industrial sites operated by WBEA from 2009 to 2017. The community sites sampled PM₂.₅ and PM₁₀, while the industrial sites sampled only PM₁₀. Sampling followed the Environment Canada National Air Pollution Surveillance (NAPS) 1-in-6-day, 24-hour schedule (Environment Canada, 2012). Samples were collected on pre-weighed 47 mm diameter ringed Teflon filters (Whatman prior to 2015; MTL from 2015 forward) using ThermoScientific Partisol 2000 U.S. EPA Federal reference method air samplers. Lab blanks and travel blanks were prepared and analyzed with each batch.

Following exposure, filters were returned to the laboratory, equilibrated in a humidity- and temperature-controlled Class 1000 clean room and re-weighed to determine net loadings of PM₂.₅ and PM₁₀ mass. Samples and blanks were then extracted in 30 mL of >18.2 MΩ-cm deionized water and analyzed for anions: fluoride (F⁻), chloride (Cl⁻), NO₃⁻, ortho-phosphate (PO₄³⁻), SO₄²⁻ and cations: NH₄⁺, sodium (Na⁺), potassium (K⁺), calcium (Ca²⁺) and magnesium (Mg²⁺) with a Dionex Model ICS-3000 IC, as described in Landis et al. (2017a).

LDs ranged from 0.001 μg m⁻³ for NH₄⁺ and Mg²⁺ to 0.010 μg m⁻³ for SO₄²⁻ and Ca²⁺. NO₃⁻ concentrations from these measurements should be considered lower limit estimates, due to potential loss
of volatile nitrates from Teflon filters (Seinfeld and Pandis, 1998; Landis et al., 2001; Edgerton et al., 2006).

2.1.3. Denuder/filter pack measurements for HNO3, NH3 and fine particulate sulfate (pSO2−), nitrate (pNO2−) and ammonium (pNH4+).

An annular denuder system (ADS) for above-canopy (i.e., 25 m agl) measurements of HNO3 and NH3 were installed at four solar-powered FHM sites (1 central, 3 remote) in mid-2013. The central site (site 1004) was ~20 km east of major point and area sources and the three remote sites were 100–125 km north (site 1007), west (site 2001) and east (site 2013) of site 1004 (see Fig. 1).

The ADS consisted of two URG Corporation (Chapel Hill, NC) 150 mm long by 30 mm diameter annular denuders in series (KCl for HNO3 and citric acid for NH3) followed by a mass flowmeter, a low-power pump and a thermostatically-controlled heater to keep the pump from seizing at low temperatures. Sample flow was 1.0–1.2 L min−1 and the system consumed <10 W and <20 W with the heater off and on, respectively. Samplers ran continuously for nominal 1 or 2 month collection periods (Aug–Nov 2013; Jun–Dec 2014; year-round from 2015 to 2017).

In January 2017, an in-line cyclone with a 2.5 μm cut-point and a 2-stage filter pack were added downstream of the denuders for measurement of pSO2−, pNO2− and pNH4+. The filter pack had a Teflon pre-filter to collect non-volatile PM and a nylon backup filter to capture volatile nitrate (U.S. EPA, 1999).

Denuders and filters were extracted in 10 mL and 30 mL, respectively, of > 18 MΩ-cm deionized water, then analyzed for SO2−, NO3− and NH4+ with a Dionex Model ICS-3000 IC following USEPA Method 10-5.0 (USEPA, 1999). LDS for monthly samples (nominal 50 m3 volume) were < 0.01 ppb for HNO3, 0.03 ppb for NH3, <0.01 μg m−3 for pSO2− and pNH4+ and 0.02 μg m−3 for pNO3− (sum of Teflon and nylon filters).

2.2. Deposition measurements and estimates

2.2.1. Ion exchange resins for bulk deposition and throughfall

Ion exchange resin collectors (IERs) were used to measure bulk deposition and throughfall deposition of inorganic sulfur and nitrogen (SO2−, NO3−, NH4+) and base cations (BC = Ca2+, K+, Mg2+ and Na+) as described by Fenn et al. (2015). Measurements of sulfur and nitrogen (S/N) deposition began in late 2008 and continued to the present. Measurements of BC covered two time periods: late 2008 through early 2010, and late 2013 to the present. Table A.2 lists the sites and associated periods of record for S/N and BC measurements.

Bulk samplers (3–5 per site) were installed in clearings adjacent to FHM sites and throughfall collectors (3–8 per site) were installed under the jack pine canopy within FHM sites. Field blanks were collected with each sample batch and used to blank-correct bulk and throughfall deposition. Sampling and analytical approaches for the TEEM program are described in detail by Fenn et al. (2015, 2018). LDS, based on 2 times the standard deviation of field blanks, were ≤0.03 kg per hectare (kg ha−1) for NH4+-N and NO3−-N, 0.08 kg ha−1 for SO2−-S and ≤ 0.04 kg ha−1 for Ca2+, Mg2+, K+ and Na+.

2.2.2. Dry deposition calculations

Dry deposition fluxes of nitrogen species (NO2−, HNO3, NH3, pNH4+ and pNO3−) were calculated using the Multi-Layer Dry Deposition Model (MLM) developed by the U.S. National Oceanic and Atmospheric Administration (MEYERS ET AL., 1998). Implementation of MLM and calculation of fluxes for calendar years 2015–2017 was performed as described by HSU et al. (2016). Hourly meteorological variables (wbem.org) for the MLM were collected at the FHM and AMS sites. This version of the MLM did not incorporate bi-directional flux for NH3, because the parameters used to estimate compensation points were not measured in the TEEM program. A recent modelling exercise (Whaley et al., 2018) investigated deposition across the AOSR with and without application of bi-directional fluxes for NH3. Results showed the bi-directional flux approach decreased NH3 deposition in some locations, but did not increase NH3 deposition anywhere in the model domain.

2.2.3. Calculation of S, N, BC and PAI deposition

Working definitions for S, N, BC and PAI deposition are shown in Equations 1–4. Throughfall SO2−-S measurements were used directly to estimate total sulfur deposition (Sdep). The sum of bulk NO3−-N and bulk NH3-N, plus modeled dry deposition of NO2−, HNO3, NH3, pNH4+ and pNO3− was used to estimate total nitrogen deposition (Ndep). Throughfall Ca2+, K+, and Mg2+ were used directly to estimate BC deposition (BCdep), while throughfall K+ and Mg2+ were adjusted downward (adjK+ and adjMg2+) to account for canopy additions (see S1 and Appendix Figure A.1), then included in the calculation of BCdep. For bulk N deposition and throughfall S deposition average annual values for the period 2009–2012 were used. For throughfall BC deposition, we used average annual values for the bracketing years 2009 and 2014. For modelled dry deposition of NO3−, HNO3 and NH3, we used the average of 2015 and 2016, and for modelled dry deposition of pNH4+ and pNO3− we used 2017 only, since this was the only year with regional PM measurements from the ADS.

Sdep = SdepW + SdepD + SdepB + SdepW + SdepH ~ SdepTF

where,

SdepW = measured or modelled wet S deposition, kg − S ha−1 a−1
SdepD = measured or modelled dry S deposition, kg − S ha−1 a−1
SdepB = measured bulk S deposition, kg − S ha−1 a−1
SdepH = measured throughfall SO2− - S deposition, kg − S ha−1 a−1
Ndep = NdepW + NdepB + NdepH + NdepD

NdepW = measured or modelled wet N deposition, kg−N ha−1 a−1
NdepB = measured or modelled dry N deposition, kg−N ha−1 a−1
NdepH = measured bulk N deposition, kg−N ha−1 a−1
NdepD = measured throughfall NO2− + NH3 deposition, kg−N ha−1 a−1

where, NdepW = measured or modelled wet N deposition, kg−N ha−1 a−1
NdepB = measured or modelled dry N deposition, kg−N ha−1 a−1
NdepH = measured bulk N deposition, kg−N ha−1 a−1
NdepD = measured throughfall NO2− + NH3 deposition, kg−N ha−1 a−1

An alternative approach for estimating Ndep that has been used by several researchers is to scale bulk nitrogen deposition based on the ratio of conservative ions, such as SO2− or Na+, in throughfall and bulk deposition (FENN ET AL., 2013; FENN AND HULTBERG, 1999). This approach is useful when measurements of nitrogenous gases and particles are unavailable; however, it assumes that total dry deposition of nitrogen species has the same spatial scale as that for SO2− or Na+. Test calculations using SO2− as the conservative tracer showed similar results to Eq. (2), but with ~10–20% higher total N deposition at sites close to oil sand operations.

BCdep = BCdepW + BCdepD + BCdepB + BCdepW + BCdepH ~ BCdepTF

where the components of BC are the four cations: Na+, Ca2+, adjK+ and adjMg2+ units are kilo-equivalents ha−1 a−1 (kg ha−1 a−1).

PAI was then calculated using Eq. (4) as the sum of acidifying components minus neutralizing components.

PAI = Sdep + Ndep − BCdep

where all are expressed as kg ha−1 a−1

This PAI calculation differs in several ways from that required by the Alberta acid deposition management framework for regulatory assessment (Government of Alberta, 2008, 2014). First, the
Alberta approach specifies a regional atmospheric deposition model to estimate Sdep, Ndep and BCdep, and we used observational data for these estimates. Second, the Alberta approach calculates PAI for a model domain of 1° by 1° grid cells covering Alberta and portions of neighboring provinces and states, while this study calculates point estimates for specific jack pine sites. Finally, the Government of Alberta uses PAI as a tool for management of acidifying emissions. The objective of the current work is to show patterns and trends across the AOSR and to provide input data for researchers studying associations between deposition and soil chemistry, foliar chemistry and plant communities (e.g., Bartels et al., 2019, this issue).

### 2.3. Statistical analyses

Data processing and statistical analyses were performed using SAS v9.4 (SAS Institute, Cary, NC) and SigmaPlot v13 (Systat Software Inc., San Jose, CA). Parametric statistics used in this analysis include least square general linear model regression, a t-test for independent samples, and one-way analysis of variance (ANOVA). The assumptions of the parametric procedures were examined using residual plots, skewness and kurtosis coefficients, Shapiro-Wilk test, and Q-Q plots. Non-parametric statistics included Kendall tau for statistical significance and Sen’s slope for changes over time. One-sided tests and a level of significance of \( p = 0.05 \) were used for all statistical procedures unless otherwise stated. Spatial interpolation plots were created in Surfer v.15 (Golden Software, Golden, CO). Data were log transformed prior kriging, and displayed as linear in the subsequent figures. All data were blank-corrected based on loadings reported for field blanks. Time-weighted annual averages were calculated for passive trace gas data and ADS data to account for variable exposure periods.

### 3. Results and discussion

#### 3.1. Spatial patterns of SO2, NO2, HNO3 and NH3

Summary statistics for SO2, NO2, HNO3 and NH3 are shown in Appendix Tables A3–A6. For SO2, long-term time-weighted average concentrations ranged from 0.30 ppb at a remote site (site 1008) to 2.76 ppb at a centrally located industrial site (MILD). Nine of 36 sites (25%) reported average concentrations <0.5 ppb and 12 sites (33%) reported average concentrations > 1.0 ppb. Long-term average NO2 concentrations ranged from 0.17 ppb at site 1991 to 5.78 ppb at MILD. Thirteen sites (36%) reported concentrations < 0.5 ppb and 16 sites (44%) reported concentrations >1.0 ppb.

Multi-year (2014–2017) average NH3 concentrations ranged from 0.19 ppb at site 2001 to 1.1 ppb at MILD. 18 of 36 sites (50%) exhibited long-term concentrations < 0.3 ppb, while 7 of 36 sites (19%) exhibited concentrations greater than 0.5 ppb. HNO3 exhibited a relatively narrow factor of 2 range in concentrations across the network (0.07 ppb at site 1995 and 0.15 ppb at site 1004). Minimum and maximum concentrations were < 0.08 ppb for 7 sites 0.12–0.15 ppb for 8 sites (22%), respectively.

Annual 2017 time-weighted average concentrations of SO2, NO2, HNO3 and NH3 are shown in Appendix Figure A2. All gases exhibit maximum concentrations near the center of oil sand operations and minimum concentrations at remote sites to the northwest and west. Data for SO2 and NO2 show a distinct southwest to northeast alignment of concentrations above 1.0 ppb. This pattern reflects mining development north of Fort McKay, but also mirrors the orientation of the Athabasca River valley. Wind direction in the valley has a strong preference for southwest and northeast and thus appears to contribute to higher concentrations in both these directions.

#### 3.2. Temporal trends in SO2, NO2, HNO3 and NH3

Regression statistics for annual time-weighted average SO2 versus year are summarized in Appendix Table A7. Non-parametric and parametric tests for trends generally yield similar slopes, but with slightly lower \( p \) values (i.e., greater number of significant results) for the parametric tests. The 18-year group show statistically significant temporal trends for 5 of 6 sites, with downward slopes on the order of −2.5% per year (roughly −0.06 ppb per year). Regression slopes imply an overall reduction in SO2 concentrations of about 35–50% over the 18-year period of record. Results are consistent with findings from Feng et al. (2019) and Tang et al. (2018) who analyzed SO2 trends in eastern Canada/U.S. and the UK, respectively. Feng et al. (2019) reported overall SO2 reductions of 88% between 1990 and 2015, while Tang et al. (2018) reported reductions of 81%. Aas et al. (2019) analyzed global and regional SO2 observations for the period 2000–2015 and reported similar downward trends for North America (−4.7% per year) and Europe (−3.9% per year), no trend for east Asia (0.1% per year) and an upward trend for Africa (5.0% per year).

The 10–12-year group shows statistically significant decreases in SO2 for 8 of 15 sites, plus borderline significance (\( p < 0.1 \)) for another 2 sites, with regression slopes from −2.0 to −11.4% per year. The 4–8-year group shows statistically significant decreasing trends for 2 out of 4 sites, with slopes ranging from −4.3% per year to −12.1% per year. Unlike the FHM sites, none of the five WBEA community/industrial sites shows a statistically significant trend for SO2.

Example time series of annual time-weighted SO2 concentrations for 4 sites (3 significant trends, 1 non-significant trend) from the 18-year group are presented in Appendix Figure A3. Concentrations in 2012 generally stand out as lower than 2013 and most years between 2014 and 2017. This is noteworthy because emission inventories indicate a ~40% decrease in SO2 emissions within the AOSR between 2012 and 2013 and stabilization at still lower levels between 2014 and 2016 (Foster et al., 2019).

Regression statistics for annual average NO2 versus year show few temporal trends (see Appendix Table A8). Within the 18-year group, 2 of 6 sites show significant trends, but with slopes of opposite sign. No significant trend is observed for sites with <18 years of data. Like SO2, a large proportion of regression slopes are negative. This suggests widespread decreases in NO2 concentrations, but insufficient statistical power at individual sites to achieve significance. Time series plots of annual NO2 concentrations for 4 example sites (2 significant upward trends, 1 significant downward trend and 1 non-significant trend) from the 18-year group are presented in Appendix Figure A4.

The absence of NO2 trends contrasts with AOSR emission inventories, which show ~25% increases in NO2 emissions from 2005 to 2010 and again from 2010 to 2015 (Foster et al., 2019). However, province-wide NO2 emissions decreased from 2000 through 2015 (Foster et al., 2019). Results for the passive NO2 network also differ from Canada-wide trends. Reid and Aherne (2016) recently analyzed data from 63 regulatory monitoring sites across Canada with at least 10 years of data. Results showed significant (~35%) overall decreases between the years 1998 and 2013, more or less in accordance with national transportation emissions reductions.

Three-year average concentration fields of SO2 and NO2, for six time periods between 2000 and 2017 (2000–2002, etc.) are shown in Fig. 2 and Fig. 3, respectively. The passive network expanded significantly from 2000 to 2017, so the figures reflect the increased number of sites, expansion of the domain, and temporal variability. During the first two triennial periods, there were two sites with average SO2 > 2 ppb and concentrations were mostly >1 ppb. The next two triennial periods showed decreased concentrations and, by 2009–11, disappearance of the 2 ppb isopleth.
Fig. 2. Spatially Interpolated 3-year average SO$_2$ concentrations (ppb) from passive monitoring sites.
Fig. 3. Spatially Interpolated 3-year average NO$_2$ concentrations (ppb) from passive monitoring sites.
The last two triennial periods show a continued decline with concentrations >1 ppb confined to 6 sites near the center of the domain, plus 1 site located to the northeast.

NO2 concentrations showed an increase at many sites for the first three time periods then a decline for the last three. The 2015–17 concentration field is similar to the 2000–03 field, but with improved definition of isopleths due to increased spatial coverage and site density.

The measurement record for HNO3 and NH3 was too short to test for annual trends; however, the data did indicate month to month and season to season variability. Monthly average concentrations of HNO3 and NH3 for the four sites with ADS are presented in Appendix Figure A.5. HNO3 data show fairly well defined seasonal differences and differences between sites from late 2013 through 2015. During this time maxima (>0.15 ppb) occurred mid-year (summer) and minima (<0.05 ppb) occurred in winter. The typical order of concentration across sites was central > north > east > west. Starting in late 2015, the month to month pattern became much less clear and maxima occurred at various times of the year. HNO3 concentrations at the central and northern still exhibited the highest concentrations in most months, but more often than not north > central.

Monthly NH3 concentrations showed more dramatic seasonality. The general pattern was one of low concentrations (<0.1 ppb) throughout the winter, an increase in early spring or about the time of snowmelt, high concentrations (≥0.25 ppb) during the summer, then a gradual decrease through the fall. Excursions with concentrations ≥ 1 ppb also occurred in different months in different years. These excursions usually involved multiple sites, but rarely all four sites. In 2016, for example, the Horse River Wildfire burned from early May through June and the single sample covering this period more than doubled the annual average concentrations. The month to month variability was observed by Li et al. (2014) for a site in rural Wyoming, USA and they attributed similar low wintertime NH3 concentrations to particulate formation.

### 3.3. Comparison with literature values of NH3, NO2, HNO3 and SO2

TEEM NH3, NO2, HNO3 and SO2 measurements are compared with literature values in Table 1. Literature and publicly available data were included in the comparison if the data record included ≥1 year of measurements, sites were located in the Northern Hemisphere and settings were described as rural, regional or remote. Data from urban sites (high NOx and photochemical products), agricultural sites (high NH3) and livestock feedlots or poultry operations (high NH3) were intentionally excluded because concentrations in these settings are known to be many times higher than those in the AOSR. The data in Table 1 are not meant to be an exhaustive compilation, but rather a basis for comparing TEEM results with those from potentially similar rural sites in the Northern Hemisphere.

The Environment and Climate Change Canada Canadian Acid Precipitation Monitoring Network (CAPMoN) measures particles and gases on a daily schedule using a three-stage open-faced filter pack (particle size cut unspecified) operated at 17 L per minute (lpm). Particles are collected on a Teflon filter, HNO3 and a portion of SO2 are collected on a nylon backup filter and the remaining SO2 is collected on carbonate-impregnated filter filters (Cheng and Zhang, 2017). The USEPA Clean Air Status and Trends Network (CASTNet) measures particles and gases on a weekly schedule.
using a similar filter pack as CAPMoN, but operated at 1.5 lpm for eastern U.S. sites and 3.0 lpm for western U.S. sites. The EU-NitroEurope and UK Acid Gas and Aerosol Network (AGANet) use similar low-flow (~0.4 lpm) denuder/filter pack systems to collect particles (size cut ~4.5 μm) and gases on monthly schedules. Tubular denuders collect HNO₃, SO₂ and NH₃, followed by a sequence of filters to collect particles plus volatilized NO₃ and NH₃ (Flechard et al., 2011; Tang et al., 2018).

Results show that concentrations of NH₃ and HNO₃ are generally consistent with those reported for background sites in Europe and North America. NH₃ data from 10 studies/networks indicate that annual average concentrations are generally ~0.2 μg m⁻³, while data from background sites in northern Europe show concentration on the order of 0.1 μg m⁻³ (Flechard et al., 2011; Fern and Hellsten, 2012). This is similar to the range observed for remote and central sites in the TEEM network (0.10–0.16 μg m⁻³).

Results for SO₂ show that concentrations in the AOSR are elevated relative to regional and remote sites in the USA and Europe. Remote sites in the TEEM network show an average concentration of ~1 μg m⁻³, which is higher than CASTNet sites in the western U.S. (range ~0.2–0.8 μg m⁻³). Interestingly, data from two CAPMoN sites in British Columbia and Saskatchewan show the same thing; however, the CAPMoN data are from 2010 and may not reflect current (2017) conditions. NO₂ datasets are sparser than the other gases, but available data suggest that concentrations at remote TEEM sites are in line with forested sites in Europe and North America.

3.4. S, N and BC in PM₁₀ and PM₂.₅

3.4.1. Spatial patterns

Summary statistics for major ions in PM₁₀ at two industrial sites and four community sites are listed in Appendix Table A.9. The sites are ordered from north to south and span roughly 105 km from SMMR to ANZC (see Fig. 1). SO₄²⁻ is the dominant anion at all sites with an average concentration of 0.86 ± 0.09 μg m⁻³ across sites and a gradient of about 30% from north to south. Ca²⁺ is the dominant cation at all sites, except ANZC, with an average concentration of 0.34 ± 0.28 μg m⁻³ across sites and a range in excess of 600%. The other major ions, NH₄⁺, K⁺ and NO₃⁻, exhibit relatively consistent concentrations across sites.

Ion balances for the six PM₁₀ monitoring sites are shown in Appendix Figure A.7. As seen for PM₁₀, Ca²⁺ and NH₄⁺ are dominant cations and SO₄²⁻ is the dominant anion. Na⁺ and Cl⁻ are minor contributors to the ion balance, except at the two sites in Fort McMurray (PATM and ATRV) where rock salt is used on roads as a de-icing agent in the winter. Comparison of the sum of measured cations (Σcations) and the sum of measured anions (Σanions) shows significant imbalances with excess Σcations at all sites. The imbalance is roughly equivalent to Ca²⁺, which suggests that much of PM₁₀ Ca²⁺ originates as CaCO₃ and that CO₃²⁻ and/or HCO₃⁻ is the missing anion. This is consistent with the use of crushed limestone for construction of mine roads and haul roads throughout the region.

The atomic composition of PM₂.₅ for the community sites (n = 4) and FHM sites (n = 4) is shown in Appendix Figure A.8A. Data for the community sites reflect 9-year averages (2009–2017), while those for FHM sites are for 2017 only. NH₄⁺ and SO₄²⁻ are the dominant cation and anion, respectively, at all eight sites. Community sites have an excess of cations over anions while FHM sites have excess anions over cations. PM₂.₅ composition therefore makes a fairly rapid transition from net cations to net anions from the river valley community sites to the surrounding forested sites.

Comparison of PM₁₀ and PM₂.₅ concentrations at the community sites shows that <5% of NH₄⁺ ~15–20% of SO₄²⁻ and >50% of Ca²⁺, Mg²⁺, Na⁺ and Cl⁻ are found in coarse PM (see Appendix Figure A.9). The near absence of NH₄⁺ in PM₁₀ suggests that most of the SO₄²⁻ is chemically associated with another cation, and may be formed heterogeneously, such as via reaction between SO₂ and alkaline PM, or during occasional fog/ice fog events in the river valley. Additional size distribution data (e.g., via multi-stage impactor) are needed to better define chemical associations and coarse particle deposition rates.

Monthly average concentrations of major ions in PM₁₀ show two distinct patterns over the course of the year (Fig. 4). NH₄⁺, SO₄²⁻ and NO₃⁻ are generally highest in winter and decrease from winter to early fall. The monthly pattern for Ca²⁺ (and Mg²⁺, not shown) is almost a mirror image of the other major ions. Lowest average concentrations occur from December through February and highest from March through August, with a factor of ~5 overall range. Landis et al. (2017a) also observed this pattern for PM samples collected at Fort McKay during 2010–2011.

Wind speed is a key variable in the suspension/resuspension of surface particulate matter (Wang et al., 2015). Meteorological data from regional sites shows that average wind speed is typically lowest from December through February and highest from March through May. The monthly pattern of Ca²⁺ and other base cations may be controlled by the combination of snow cover/frozen ground and low wind speeds in winter, followed by thawing/drying of soils and higher wind speeds in spring and summer.

Linear regressions of annual average PM₁₀ components versus year show mostly non-significant (p ~0.1–0.3) trends for 2009–2017 (Appendix Table A.10). Exceptions to this include NH₄⁺ (significant negative slope at all sites except PATM), and SO₄²⁻ (significant negative slope at ANZC only). Similar to SO₂ and NO₃⁻, most slopes are negative and correspond to concentration decreases of a few % per year; however, the data record is insufficient determination statistically significant trends. Aas et al. (2019) noted similar, but statistically significant, rates of decline for aerosol sulfate over Europe and North America from 2000-2015. Feng et al. (2019) reported declines in SO₄²⁻, NO₃⁻ and NH₄⁺ of 73%, 29% and 66%, respectively, across eastern Canada/U.S. from 1990 to 2015; and Tang et al. (2018) reported declines in SO₄²⁻, NO₃⁻ and NH₄⁺ of 69%, 52% and 62%, respectively, across the UK from 1999 to 2015. The spatial distributions of major ions in PM₂.₅ during 2017 are shown in Appendix Figure A.10. In general, SO₄²⁻, NO₃⁻ and NH₄⁺ exhibit ranges of about a factor of 2–3, and concentrations are highest at the three community sites in the Athabasca River valley (BFGM, PATM and ATRV) and lowest at the FHM sites 100–125 km to the east and west. Ca²⁺ concentrations show a broader range (factor of 12+) and are much higher at the community sites. Small differences between the westernmost, northernmost and easternmost forest health sites may reflect upwind/downwind effects surrounding the oil sand production operations.

3.4.2. Comparison with literature values for NH₄⁺, NO₃⁻ and non-sea salt SO₄²⁻

TEEM data for PM₂.₅ NH₄⁺, NO₃⁻ and non-sea salt SO₄²⁻ are compared with literature values in Table 2. nssSO₄²⁻ is used instead of measured SO₄²⁻ to avoid positive biases at coastal or near-coastal sites with high concentrations of sea salt PM (SO₄²⁻ via Na⁺ in bulk seawater ~0.25) (Keene et al., 1986). The difference between nssSO₄²⁻ and SO₄²⁻ at TEEM sites, and many inland sites, is ~2%, but differences in the U.K., for example, have been shown to be as high as 83% (Tang et al., 2018). Also shown in Table 2 are t-NH₄⁺ (i.e., NH₄⁺ + NH₃ expressed as NH₄⁺) and t-NO₃⁻ (HNO₃ and NO₃⁻ expressed as NO₃⁻). These are included, where possible, because totals may be more representative when comparing data across several different techniques. Measurement approaches for CAPMoN, CASTNet and AGANet were described previously. For the U.S. Interagency Monitoring of Protected Visual Environments...
(IMPROVE) network, PM$_{2.5}$ samples (following removal of HNO$_3$ and SO$_2$) are collected on nylon filters every third day and analyzed for SO$_4^{2-}$ and NO$_3^-$ (Malm et al., 1994).

NH$_4^+$ and NO$_3^-$ concentrations show roughly a factor of 10 range within each network (factor of 6 for CAPMoN), with minimum values ~0.1–0.2 μg m$^{-3}$ and maximum values ~1–2 μg m$^{-3}$. The two

![Graphs showing monthly mean PM$_{10}$ NH$_4^+$, Ca$^{2+}$, SO$_4^{2-}$ and NO$_3^-$ at an industrial site (MSRV) and three community sites (BGFM, ATHV and ANZC).]
exceptions to this are Europe with a maximum NO$_3$ concentration of 4.4 µg m$^{-3}$ and the UK with a minimum NO$_3$ concentration of 0.4 µg m$^{-3}$. Results for the AOSR show that NH$_4^+$ and NO$_3$ at remote sites correspond to the low end of observations from CA, USA, UK and Europe, while those from central and community sites vary by almost a factor of 20 across sites. Mean SO$_4^{2-}$ concentration is 20–14 months of deposition. Mean bulk NH$_4^+$ loadings of SO$_4^{2-}$ are observed at 3 central sites and 9 remote sites, respectively. Throughfall loadings of SO$_4^{2-}$ are 2–3 times higher.

<table>
<thead>
<tr>
<th>Location (Program)</th>
<th>Meas. Period</th>
<th>Method</th>
<th>PM fraction</th>
<th>Setting</th>
<th>NH$_4^+$ (µg m$^{-3}$)</th>
<th>NO$_3$ (µg m$^{-3}$)</th>
<th>nssSO$_4^{2-}$ (µg m$^{-3}$)</th>
<th>t-NH$_4^+$ (µg m$^{-3}$)</th>
<th>t-NO$_3$ (µg m$^{-3}$)</th>
<th>Ref.</th>
</tr>
</thead>
<tbody>
<tr>
<td>AOSR, CA (WBEA-TEEM)</td>
<td>2017</td>
<td>TF</td>
<td>2.5</td>
<td>community (n = 4)</td>
<td>0.20–0.34</td>
<td>0.07–0.19</td>
<td>0.54–0.88</td>
<td>n.m.</td>
<td>n.m.</td>
<td>this study</td>
</tr>
<tr>
<td>AOSR, CA</td>
<td>2017</td>
<td>FP</td>
<td>–</td>
<td>remote (n = 3)</td>
<td>0.09–0.13</td>
<td>0.07–0.14</td>
<td>0.32–0.41</td>
<td>0.14–0.24</td>
<td>0.24–0.65</td>
<td></td>
</tr>
<tr>
<td>AOSR, CA</td>
<td></td>
<td>–</td>
<td>–</td>
<td>central (n = 1)</td>
<td>0.20</td>
<td>0.14</td>
<td>0.57</td>
<td>0.39</td>
<td>0.61</td>
<td></td>
</tr>
<tr>
<td>Saturna, BC, CA</td>
<td>2010</td>
<td>FP</td>
<td>–</td>
<td>open</td>
<td>0.22</td>
<td>0.55</td>
<td>0.64</td>
<td>n.m.</td>
<td>1.2</td>
<td>CAPMoN, 2019</td>
</tr>
<tr>
<td>Bratt’s Lake, SK, CA</td>
<td></td>
<td></td>
<td>–</td>
<td>–</td>
<td>0.56</td>
<td>0.97</td>
<td>1.1</td>
<td>n.m.</td>
<td>1.5</td>
<td></td>
</tr>
<tr>
<td>E. CA</td>
<td></td>
<td></td>
<td>–</td>
<td>–</td>
<td>0.19–1.1</td>
<td>0.13–1.9</td>
<td>0.64–2.1</td>
<td>n.m.</td>
<td>0.23–2.7</td>
<td></td>
</tr>
<tr>
<td>E. USA (USEPA-CASTNet)</td>
<td>2017</td>
<td>FP</td>
<td></td>
<td>open</td>
<td>0.18–0.75</td>
<td>0.18–1.5</td>
<td>0.50–1.7</td>
<td>n.m.</td>
<td>0.37–2.4</td>
<td>CASTNet, 2019</td>
</tr>
<tr>
<td>W. USA</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>0.12–0.54</td>
<td>0.10–1.2</td>
<td>0.27–1.4</td>
<td>n.m.</td>
<td>0.25–2.5</td>
<td></td>
</tr>
<tr>
<td>Alaska, USA</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>0.09</td>
<td>0.05</td>
<td>0.27</td>
<td>n.m.</td>
<td>0.11</td>
<td></td>
</tr>
<tr>
<td>E. USA (IMPROVE)</td>
<td>2017</td>
<td>NF</td>
<td>2.5</td>
<td>regional (n = 32)</td>
<td>n.m.</td>
<td>0.13–1.2</td>
<td>0.52–1.3</td>
<td>n.m.</td>
<td>n.m.</td>
<td>IMPROVE, 2019</td>
</tr>
<tr>
<td>W. USA (*)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>n.m.</td>
<td>0.06–1.0</td>
<td>0.22–1.3</td>
<td>n.m.</td>
<td>n.m.</td>
<td></td>
</tr>
<tr>
<td>Alaska, USA (*)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>n.m.</td>
<td>0.04–0.09</td>
<td>0.24–0.47</td>
<td>n.m.</td>
<td>n.m.</td>
<td></td>
</tr>
<tr>
<td>Europe (EU-Nitro Europe IP)</td>
<td>2007–08</td>
<td>FP</td>
<td>–</td>
<td>forest (n = 29)</td>
<td>0.15–1.7</td>
<td>0.09–4.4</td>
<td>0.32–6.7</td>
<td>0.27–5.9</td>
<td>Flechard et al., 2011</td>
<td></td>
</tr>
<tr>
<td>UK (UK-AGANET)</td>
<td>2015</td>
<td>FP</td>
<td>–</td>
<td>rural bgnd. (n = 28)</td>
<td>0.13–0.94</td>
<td>0.40–1.8</td>
<td>0.10–0.50</td>
<td>0.20–6.8</td>
<td>0.48–2.3</td>
<td>AGANET, 2019</td>
</tr>
</tbody>
</table>

TF = Teflon filter; FP = Teflon + Nylon filters; NF = nylon filter; n.m. = not measured
IMPROVE = Interagency Monitoring of Protected Visual Environments

5.3. Deposition of S, N and BC

5.3.1. IER deposition data

Annualized (2011–2012) bulk NH$_4^+$/N, bulk NO$_3^-$/N and throughfall SO$_4^{2-}$/S deposition are summarized in Appendix Table A.11 for 20 sites with ≥3 years of data. A valid year is defined herein as one that includes consecutive warm and cold seasons that together represent 10–14 months of deposition. Mean bulk NH$_4^+$ deposition varies across the domain by almost a factor of 3. Mean loadings ≥1.2 kg-N ha$^{-1}$ a$^{-1}$ are observed at 5 sites located near oil sand operations as a major source of Ca$^{2+}$ and other crustal elements.

Increases in deposition of Ca$^{2+}$ are plotted as a function of distance to the nearest stack, while Ca$^{2+}$ is plotted as a function of distance to the center of the nearest mining operation. The latter approach is based on findings of Landis et al. (2017b, 2019) who identified surface oil sand production operations as a major source of Ca$^{2+}$ and other crustal elements in PM collected at Fort McKay and in lichen samples collected throughout the region. Steep declines in deposition with increasing distance are apparent for all four components, and a reasonable fit can be produced with a simple inverse-distance regression curve. These inverse-distance curves suggest three spatial domains surrounding oil sand activities: a “source-dominated” domain close to mining operations, a “transition” domain extending out to 40–80 km, and a “background” dominated domain further afield. This is consistent with lichen biomonitoring results (Landis et al., 2019), but is uncertain because there are no throughfall measurements for BC between 12 km and 50 km, and only a single measurement for S and N in this range.
3.5.2. Ndep Sdep, BCdep and PAI at FHM sites

Estimated Ndep is shown by site and component in Fig. 6, with sites arranged from west to east to provide a cross-section of the AOSR. Ndep ranges from 5.7 kg-N ha\(^{-1}\)a\(^{-1}\) at site 2012 to slightly < 2 kg-N ha\(^{-1}\)a\(^{-1}\) at several remote sites. Total deposition exceeds 3 kg-N ha\(^{-1}\)a\(^{-1}\) at 10 sites and 4 kg-N ha\(^{-1}\)a\(^{-1}\) at 5 central sites. Except for dry NO\(_2\)-N, the percent contribution of individual components to Ndep varies by about a factor of 2 across the region: bulk NH\(_4\)-N (24–41%), bulk NO\(_3\)-N (17–24%), dry NO\(_2\)-N (8–42%), dry HNO\(_3\)-N (8–17%), dry NH\(_2\)-N (5–12%), dry pNO\(_3\)-N (2–4%) and dry pNH\(_4\)-N (1–3%). For the centrally located sites, deposition estimates rank as follows: dry NO\(_2\)-N > bulk NH\(_4\)-N > bulk NO\(_3\)-N > dry NH\(_2\)-N > dry HNO\(_3\)-N > dry pNO\(_3\)-N > dry pNH\(_4\)-N. At the most distant sites the order of contributions was: bulk NH\(_4\)-N > bulk NO\(_3\)-N > dry HNO\(_3\)-N > dry NO\(_2\)-N > dry NH\(_2\)-N > dry pNO\(_3\)-N > dry pNH\(_4\)-N.

Dry deposition of pNO\(_3\)-N is underestimated due to lack of PM\(_{10}\) data at regional sites. Zhang et al. (2009) measured size distributions of NO\(_3\) at eight sites across southeastern Canada and showed that 10–77% resided in the > PM\(_{2.5}\) size fraction, and varied as a function of site characteristics (e.g., polluted versus remote) and season. The contribution of pNO\(_3\)-N deserves further investigation, but an underestimate of a factor of 2 would not appreciably alter total deposition estimates nor would it affect the above rankings of components. This issue does not arise with dry pNH\(_4\)-N because almost all NH\(_4\)\(^+\) is found in the PM\(_{2.5}\) fraction.

Reduced N (i.e., dry NH\(_3\) deposition plus bulk NH\(_4\)-N deposition) accounts for 28–58% of Ndep and this percentage generally increases from central sites (i.e., where dry NO\(_2\) dominates) to remote sites. However, Fenn et al. (2015) noted that reduced N deposition significantly exceeded oxidized N deposition at industrial sites very close to operations. Bulk NH\(_4\) and bulk NO\(_3\) together represent ~40–62% of Ndep and this percentage increases with distance to operations. In other words, wet and dry processes contribute more or less equally to Ndep in the AOSR.

The current Ndep estimates for FHM sites are roughly half of those reported by Zhang et al. (2009) for eight sites in a west-east transect across Canada from Chalk River, Ontario to Kejimkujik, Nova Scotia (4–10 kg-N ha\(^{-1}\)a\(^{-1}\) versus <2–6 kg-N ha\(^{-1}\)a\(^{-1}\) ). However, the relative contribution of components followed a similar order with wet NH\(_4\)-N > wet NO\(_3\)-N > NO\(_2\) ~ HNO\(_3\) > NH\(_3\) > pNO\(_3\), pNH\(_4\)-N.

The current Ndep estimates do not include organic forms of nitrogen (ON) that are known to occur in precipitation, gases and PM (Cornell, 2011). Cape et al. (2012) measured ON in wet deposition at 18 sites across Europe and found that ON deposition ranged from 0.15 to 1.74 kg-N ha\(^{-1}\)a\(^{-1}\) and contributed 2–36% (median 10%) of dissolved N deposition on an annual basis. Zhang et al. (2009) measured gas phase oxidized ON and estimated dry deposition rates for ON at eight sites across southeastern Canada. Results showed ON deposition to be ≤ 0.75 kg-N ha\(^{-1}\)a\(^{-1}\) for “polluted” sites, ≤0.25 kg-N ha\(^{-1}\)a\(^{-1}\) for “less polluted” sites and ≤ 0.1 kg-N

![Graphs showing deposition vs. distance curves for bulk NH\(_4\)-N, bulk NO\(_3\)-N, throughfall SO\(_4\)-S and throughfall Ca\(^{2+}\). Dashed lines represent 95% confidence intervals.](image)
Data also suggest that ON can represent a significant fraction (10–33%) of total N in PM$_{2.5}$ (Zhang et al., 2002; Lin et al., 2010; Rastogi, et al., 2011; Matsumoto et al., 2019). If aerosol ON is primarily in the PM$_{2.5}$ fraction, then the contribution of ON dry deposition to Ndep is likely <5%. However, little is known about ON in larger size fractions, and therefore particulate ON deposition could be important. Further research is needed to understand atmospheric concentrations, sources and effects of ON in the AOSR.

Comparison of Ndep and throughfall N deposition (Ndep$_{TF}$) for 9 sites with Ndep in the range of 2 to 6 kg N ha$^{-1}$ a$^{-1}$ shows that Ndep$_{TF}$ accounts for about 42%–85% of Ndep (see Appendix Figure A.12) and clearly indicates that canopy consumption of atmospheric N is affecting throughfall measurements. The difference between the two fluxes suggests nearly constant canopy uptake of 1.1 ± 0.5 kg N ha$^{-1}$ a$^{-1}$. If canopy uptake of atmospheric N is constant across the AOSR deposition gradient, then Ndep$_{TF}$ or an adjusted form of Ndep$_{TF}$ may serve as a useful surrogate for

Fig. 6. Estimated 2011–2012 components of Total-Nitrogen (Total-N) deposition to TEEM sites. Absolute deposition in upper panel and relative contributions in lower panel. Sites are arranged west to east on the x-axis.
Ndep in areas of high deposition or at sites lacking inputs to drive dry deposition calculations. This is analogous to the treatment of throughfall K\(^+\) and Mg\(^{2+}\) for BC estimates.

Estimated Sdep and BCdep are displayed by site in Fig. 7. The pattern for Sdep is similar to Ndep, with maximum deposition near the center of the domain, but with a broader range of deposition. Sdep is \(<2\) kg-S ha\(^{-1}\) a\(^{-1}\) at 3 remote sites and \(>6\) kg-S ha\(^{-1}\) a\(^{-1}\) at 10 central sites. Ca\(^{2+}\) is the dominant BC, followed by adjusted K\(^+\), adjusted Mg\(^{2+}\), and Na\(^{+}\). In contrast to Ndep and Sdep, BC shows two peaks of elevated deposition. One of the peaks coincides with maximum Sdep and Ndep, and the second peak occurs east and north of the first and includes the three sites with highest BC deposition. Individual components of deposition at FHM sites are listed in Appendix Table A.7.

Estimated PAI for FHM sites is mostly in the range of 0.1–0.2 k eq ha\(^{-1}\) a\(^{-1}\) (see Fig. 8). The extreme western edge of the domain

![Fig. 7. Estimated 2011–2012 deposition of Total-Sulfur (above) and individual base cations (below). Sites are arranged west to east on the x-axis.](image-url)
exhibits PAI values of 0–0.1 keq ha$^{-1}$ a$^{-1}$, and there are two distinct areas of significantly positive and zero to negative PAI. The area of highest PAI includes values of 0.3 to 0.7 keq ha$^{-1}$ a$^{-1}$. It is centered slightly east of the center of oil sands operations and extends roughly 50 km north–south and 25 km east–west. The area of lowest PAI includes values of 0.0 to $-0.5$ keq ha$^{-1}$ a$^{-1}$ and is located 50 km northeast of the center of operations. The overall pattern shows slightly positive (acidifying) values in those parts of the domain with low Sdep, Ndep and BCdep, and much more positive values near the center of the domain with highest Sdep, Ndep and BCdep. Negative or near-zero PAI values are suggested where there are sources of BC (mines and other surface operations) in areas with moderate to low Sdep and Ndep. Watmaugh et al. (2014) also demonstrated the importance of fugitive dust emissions and BCdep for mitigating acidification across the AOSR.

PAI values for most of the domain are below critical loads for sensitive soils that have been established by the Government of Alberta (Government of Alberta, 2008); that is < 0.25 keq ha$^{-1}$ a$^{-1}$. Makar et al. (2018) recently estimated that up to 10% of Alberta and portions of Saskatchewan downwind of oil sand production operations exceed critical loads. However, the Makar et al. (2018) analysis was based on significantly different methods for calculating both atmospheric deposition (atmospheric model versus observation-based) and soil sensitivity. Thus, there is a need for further efforts to reconcile approaches for estimating deposition and associated effects across the AOSR. Results of this work will be used in two companion papers that assess impacts of Sdep, Ndep and PAI on soils and vegetation at FHM sites (Bartels et al., 2019; MacKenzie and Dietrich, 2019).

3.5.3. Comparison of TEEM deposition estimates with literature values

TEEM deposition estimates are compared with literature values and publicly available datasets in Table 3. Results must be compared with caution for a number of reasons, including differences in precipitation type and amount across regions and differences in the approach used to estimate dry deposition. For example, annual precipitation in the AOSR is in the range of 400–450 mm per year, which is roughly 50% of that over much of eastern Canada and the U.S., but 100–300% higher than arid parts of the southwestern U.S. Given this range in precipitation, highly variable deposition rates are not surprising.

Results show that bulk deposition at remote TEEM sites is similar to that in the eastern US, eastern CA and NW Europe, but higher than the lowest deposition rates in western CA and US. Bulk NH$_4^+$ and NO$_3^-$ deposition at central sites is closer to the high end observed for western CA and US, while bulk nss-S is as high or higher than eastern CA and US. For total deposition, remote sites fall towards the low end of reported values for N, nss-S and Ca$^{2+}$, and central sites fall towards the high end of reported deposition estimates.

Fig. 8. Estimated 2011–2012 potential acid input (PAI) for TEEM monitoring sites.
### Table 3

Comparison of TEEM deposition with literature values.

<table>
<thead>
<tr>
<th>Location (Program)</th>
<th>Method</th>
<th>Setting</th>
<th>Total kg ha⁻¹ a⁻¹</th>
<th>N</th>
<th>NH₄⁻N</th>
<th>NO₃⁻N</th>
<th>HNO₃</th>
<th>NH₃</th>
<th>PM₁₀⁻⁴</th>
<th>PM₁₀⁻⁵</th>
</tr>
</thead>
<tbody>
<tr>
<td>AOSR, CA (WBEA-TEEM)</td>
<td>2009–12</td>
<td>RI</td>
<td>Bulk</td>
<td>0.8–1.0</td>
<td>0.6–1.2</td>
<td>0.6–1.0</td>
<td>0.4–0.7</td>
<td>n.r.</td>
<td>1.9–2.1</td>
<td>1.0–1.5</td>
</tr>
<tr>
<td>WBEA TEEM</td>
<td>2009–12</td>
<td>B/I remote (n = 6)</td>
<td>Bulk</td>
<td>0.8–1.0</td>
<td>0.6–1.2</td>
<td>0.6–1.0</td>
<td>0.4–0.7</td>
<td>n.r.</td>
<td>1.9–2.1</td>
<td>1.0–1.5</td>
</tr>
<tr>
<td>Canada 2009–12</td>
<td>B/I central (n = 3)</td>
<td>Bulk</td>
<td>1.3</td>
<td>2.0</td>
<td>1.0</td>
<td>0.6–1.3</td>
<td>n.r.</td>
<td>4.6–5.8</td>
<td>1.0–1.5</td>
<td></td>
</tr>
<tr>
<td>E. Canada 2009–12</td>
<td>Rural-Ag. (n = 7)</td>
<td>Bulk</td>
<td>0.15–1.5</td>
<td>0.2–1.1</td>
<td>0.2–0.5</td>
<td>0.1–1.0</td>
<td>n.r.</td>
<td>0.6</td>
<td>1.0–1.5</td>
<td></td>
</tr>
<tr>
<td>Aherne, 2012</td>
<td>W. USA (USEPA-CASTNet)</td>
<td>2009–12</td>
<td>Bulk</td>
<td>0.1–3.1</td>
<td>0.1–1.7</td>
<td>0.1–2.1</td>
<td>0.1–4.0</td>
<td>1.9–11</td>
<td>0.4–2.6</td>
<td>0.5–11</td>
</tr>
<tr>
<td>Benedict et al., 2013</td>
<td>NE. Europe (EMEP)</td>
<td>2009–12</td>
<td>Bulk</td>
<td>1.0–3.5</td>
<td>1.0–3.1</td>
<td>1.0–2.0</td>
<td>n.r.</td>
<td>n.r.</td>
<td>n.r.</td>
<td>n.r.</td>
</tr>
<tr>
<td>EMEP, 2019</td>
<td>China 2006–2014</td>
<td>Bulk</td>
<td>1.00</td>
<td>0.50</td>
<td>n.r.</td>
<td>n.r.</td>
<td>2.9</td>
<td>n.r.</td>
<td>n.r.</td>
<td></td>
</tr>
</tbody>
</table>

B = bulk precipitation; W = wet-only precipitation, I = inferred dry with measured gases/particles; M = modeled dry; n.r. = not reported. EMEP = European Monitoring and Evaluation Programme.

#### 4. Conclusions

Passive data from air and deposition networks operated by the WBEA TEEM program provide useful data for evaluation of spatial variability and temporal trends, and estimates of S, N, BC and PAI deposition to FHM jack pine sites.

Ambient SO₂ and NO₂ show large (factor of 5–20) spatial gradients with highest concentrations near oil sand operations and lowest concentration at the extremes of the network. Annual average SO₂ concentrations decreased significantly, and more or less linearly, across the AOSR since 2000. NO₂ concentrations showed few significant trends, but generally increased from 2000 to 2009 or 2010, then returned to near-2000 levels by 2017. HNO₃ concentrations vary by a factor of 2–3 across the domain and have been relatively consistent over time. NH₄⁺ concentrations vary by roughly a factor of 6, and show dramatic, site-specific influence from the 2016 Horse River wild fire, and wild fires in previous years. Concentrations of NH₃ and HNO₃ at TEEM sites are near the low end of those reported by networks in Canada, the U.S. and Europe. In contrast, SO₂ concentrations at TEEM sites are higher than rural/remote sites in the western U.S. and similar to sites in the eastern U.S. and eastern Canada.

PM₁₀ and PM₁₀ data are relatively limited and additional sites upwind and downwind of sources are needed to understand spatial patterns. The coarse fraction (PM₁₀⁻⁴) at industrial and community sites accounts for a large percentage of BC and ion balances show significant alkalinity. The fine fraction (PM₁₀⁻⁵) accounts for the majority of NH₄⁺, SO₄²⁻ and NO₃⁻ at community sites. Ion balances for PM₁₀ show excess cations at community sites, but exceed anions at FHMM sites. PM₁₀ data show striking seasonality with highest NH₄⁺ and SO₄²⁻ and NO₃⁻ during the winter and highest Ca²⁺ during spring and summer. Like NH₃, surface emissions of Ca²⁺ may be considerably reduced by snow cover and frozen ground. Local limestone quarrying and crushing operations are also suspended during the winter, thereby temporarily eliminating a potentially significant source of Ca²⁺. Trend analyses suggest that most components of PM₁₀ are declining at WBEA sites, but the only significant (p < 0.05) trends are for NH₃ (all sites) and SO₄²⁻ (1 site). PM₁₀ concentrations of NH₄⁺ and NO₃⁻ at remote TEEM sites are similar to the lowest concentrations reported for Canada, the U.S. and Europe.

Canopy uptake reduces throughfall deposition of NH₄⁺ and NO₃⁻ to FHM jack pine sites by about 1 kg-N ha⁻¹ a⁻¹. As a result, bulk deposition is combined with modelled dry deposition of nitrogenous gases and particles to obtain estimates of Ndep. In contrast, the jack pine canopy contributes significantly to throughfall deposition of K⁺ and Mg²⁺ and measured loadings must be adjusted downward for calculating BCdep and PAI. Further research is needed to better understand canopy exchanges in TEEM jack pine stands and to guide interpretation of throughfall results. The coarse fraction (PM₁₀⁻⁴) at industrial and community sites accounts for a large percentage of BC and ion balances show significant alkalinity. The fine fraction (PM₁₀⁻⁵) accounts for the majority of NH₄⁺, SO₄²⁻ and NO₃⁻ at community sites. Ion balances for PM₁₀ show excess cations at community sites, but excess anions at FHMM sites. PM₁₀ data show striking seasonality with highest NH₄⁺ and SO₄²⁻ and NO₃⁻ during the winter and highest Ca²⁺ during spring and summer. Like NH₃, surface emissions of Ca²⁺ may be considerably reduced by snow cover and frozen ground. Local limestone quarrying and crushing operations are also suspended during the winter, thereby temporarily eliminating a potentially significant source of Ca²⁺. Trend analyses suggest that most components of PM₁₀ are declining at WBEA sites, but the only significant (p < 0.05) trends are for NH₃ (all sites) and SO₄²⁻ (1 site). PM₁₀ concentrations of NH₄⁺ and NO₃⁻ at remote TEEM sites are similar to the lowest concentrations reported for Canada, the U.S. and Europe.

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Estimated Ndep and Sdep at FHMM sites were ~2.0–5.7 kg-N ha⁻¹ a⁻¹ and ~2.0–14 kg-S ha⁻¹ a⁻¹, respectively. Bulk deposition of NH₄⁻N and NO₃⁻N accounted for >40% of total-N deposition networkwide. Dry deposition calculations showed that NO₂ was dominant at sites close to operations, but that HNO₃, NO₂ and to a lesser extent NH₃, contributed similarly to dry deposition at remote sites. Dry NH₃ deposition and dry HNO₃ deposition were as least a factor of 2 lower than previously estimated for the AOSR (Hsu et al., 2016). This can be explained by changes in the passive sampler collection substrates, blank-correction and adjustments that were performed to correct a previous calculation error (NH₃ only).
Deposition gradients around sources were reasonably well explained ($r^2 > 0.4$) by a simple inverse distance relationship between receptor sites and sources. Sdep, Ndep and BCdep at near field sites within 10–15 km of sources were high and variable. Sdep, Ndep and BCdep at remote sites ≥ 50 km from sources were low and relatively constant, suggesting that deposition at this distance approached regional values.

PAI estimates showed net acidic deposition for 2011–2012 to be in the range of 0.1–0.2 keq ha$^{-1}$ a$^{-1}$ for the majority of sites. Higher values (0.3–0.7 keq ha$^{-1}$ a$^{-1}$) were predicted near the center of oil sand operations, while negative values (net alkalinity) were predicted in an area of relatively new surface mining activity northeast of Fort McKay. These negative PAI values are driven by a large extent by the spatial interpolation approach used to estimate BC deposition. Results are consistent with spatial patterns for lichen biomonitoring samples collected between 2008 and 2017 (Landis et al., 2019), but should be confirmed by actual measurements.

Declaration of Competing Interest

The authors declare that there is no conflict of interest regarding the publication of this article.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.scitotenv.2019.134864.

References

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